SPOTLIGHTS ...

Polyoxometalates

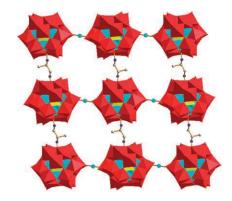
S.-T. Zheng, M.-H. Wang, G.-Y. Yang*

Extended Architectures Constructed from Sandwich Tetra-Metal-Substituted Polyoxotungstates and Transition-Metal Complexes

Chem. Asian J.

DOI: 10.1002/asia.200700166

All joined up: Hydrothermal synthesis produces three unprecedented coordination polymers constructed from sandwich tetra-transition-metal-substituted polyoxotungstates and transition-metal complexes. These compounds were structurally characterized by X-ray diffraction analysis and exhibit reversible water-sorption capabilities.



Biosynthesis

A. C. Stewart, I. J. Clifton, R. M. Adlington, J. E. Baldwin,* P. J. Rutledge*

A Cyclobutanone Analogue Mimics Penicillin in Binding to Isopenicillin N Synthase

ChemBioChem

DOI: 10.1002/cbic.200700176

Switching β-lactam for cyclobutanone

creates a new class of penicillin-like ring system, in which the β -lactam nitrogen atom is replaced by carbon. But how similar really is the structure of this new carbocyclic ring system to a penicillin? By cocrystallising the cyclobutanone analogue of isopenicillin N with isopenicillin N synthase (the central enzyme in penicillin biosynthesis), we have addressed this question.

Ionic Liquids

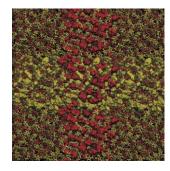
T. Köddermann, D. Paschek, R. Ludwig*

Molecular Dynamic Simulations of Ionic Liquids: A Reliable Description of Structure, Thermodynamics and Dynamics

ChemPhysChem

DOI: 10.1002/cphc.200700552

Move with the force: A new force field for simulating imidazolium-based ionic liquids [C_nmim][NTf₂] (see figure) reproduces with good agreement the experimental values for the heats of vaporization, shear viscosities and NMR rotational correlation times. The results suggest that the inaccuracies arising from previous methods are due to the overestimation of the Lennard–Jones interactions.



Structural Biology

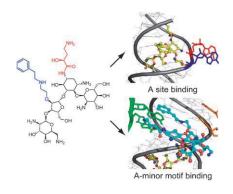
J. Kondo, K. Pachamuthu, B. François, J. Szychowski,

S. Hanessian,* E. Westhof*

Crystal Structure of the Bacterial Ribosomal Decoding Site Complexed with a Synthetic Doubly Functionalized Paromomycin Derivative: a New Specific Binding Mode to an A-Minor Motif Enhances in vitro Antibacterial Activity

ChemMedChem

DOI: 10.1002/cmdc.200700113



The synthetic paromomycin analogue with the L-haba group and an ether chain with an *O*-phenethylaminoethyl group could specifically bind to ribosomes in two different modes: 1) the classical binding to the A site and 2) binding to an A-minor motif participating in the recognition of the codonanticodon helix or in the intersubunit bridges.

... ON OUR SISTER JOURNALS



The title heterocycles are prepared and electrochemically and chemically reduced to their long-lived radical anions, which can be isolated in the form of the thermally stable salts of the $[K(18\text{-crown-6})]^+$ cation. These salts reveal low-temperature antiferromagnetic ordering of the spin system $(J = -1.65 \text{ cm}^{-1})$. Color code: gray = C, blue = N, yellow = S, magenta = Se.



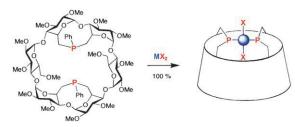
Radical Anion Salts

I. Yu. Bagryanskaya, Yu. V. Gatilov, N. P. Gritsan,* V. N. Ikorskii, I. G. Irtegova, A. V. Lonchakov, E. Lork, R. Mews,* V. I. Ovcharenko,* N. A. Semenov, N. V. Vasilieva, A. V. Zibarev*

[1,2,5]Selenadiazolo[3,4-c]-[1,2,5]thiadiazole and [1,2,5]Selenadiazolo[3,4-c][1,2,5]thiadiazolidyl – A Synthetic, Structural, and Theoretical Study

Eur. J. Inorg. Chem.

DOI: 10.1002/ejic.200700501



Tailor-made for *trans* **chelation:** A remarkably rigid diphosphane has been obtained by capping the entrance of a cyclodextrin with two "PPh" units. When treated with Group 10 metal

halides, this ligand not only behaves exclusively as a chelating ligand, but also selectively forms *trans* complexes (see picture).

Phosphane Ligands

L. Poorters, D. Armspach,* D. Matt,* L. Toupet, S. Choua, P. Turek

Synthesis and Properties of TRANSDIP, a Rigid Chelator Built upon a Cyclodextrin Cavity: Is TRANSDIP an Authentic *trans*-Spanning Ligand?

Chem. Eur. J.

DOI: 10.1002/chem.200700831

A fluoride sandwich: A cylindrophane macrocycle framed with cyanuric acid rings selectively complexes fluoride ion by a combination of anion- π interactions and ion-pair-reinforced hydrogen bonds. This compound is the first example of a purpose-designed host that exploits anion- π bonding.



Anion-π Interactions

M. Mascal,* I. Yakovlev, E. B. Nikitin, J. C. Fettinger

Fluoride-Selective Host Based on Anion– π Interactions, Ion Pairing, and Hydrogen Bonding: Synthesis and Fluoride-Ion Sandwich Complex

Angew. Chem. Int. Ed. DOI: 10.1002/anie.200704005



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